

AN EFFICIENT ENANTIOSELECTIVE PREPARATION OF 2-SUBSTITUTED-
3-HYDROXYPROPIONIC ACIDS VIA CHEMO-ENZYMATIC REACTION

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Summary: The key intermediates, 2-substituted-3-hydroxypropionic acids **1**, of the potent renin inhibitors were synthesized enantioselectively starting from 2-substituted-1,3-propanediols via chemo-enzymatic reaction.

Recently we reported¹ a novel renin inhibitor, BW-175, which is a non-peptidic and orally active low molecular inhibitor having (2*S*)-3-ethylsulfonyl-2-(1-naphthylmethyl)propionic acid.² Its benzyl analogue at position-2, BW-262, is also a potent renin inhibitor. Our strategy for the synthesis of enantiomerically pure 2-substituted-3-alkyl(or aryl)sulfonylpropionic acids is based on retro synthetic analysis shown in Fig. 1. 2-Substituted-3-hydroxypropionic acids **1** were chosen as the versatile key intermediates. The optically active 3-hydroxypropionic acid derivatives are fundamental chiral building blocks. However only a few approaches to their preparation have been reported.³ In this paper we report the enantioselective synthesis of 2-substituted-3-hydroxypropionic acids **1** from 2-substituted-1,3-propanediols **2** via lipase-catalyzed reaction.⁴

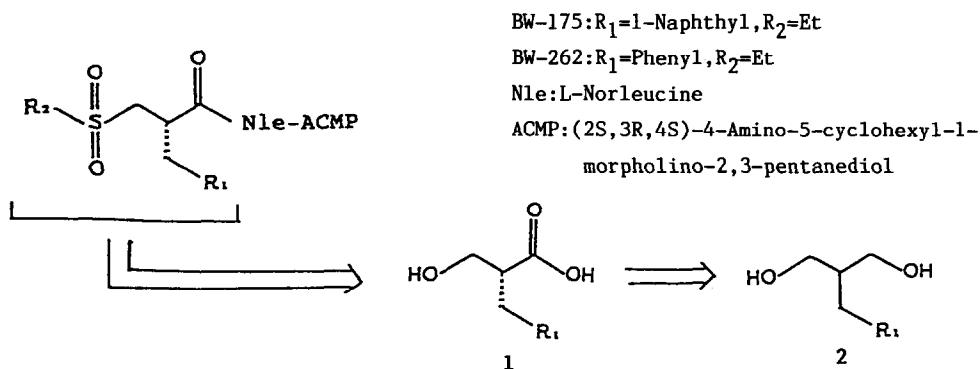
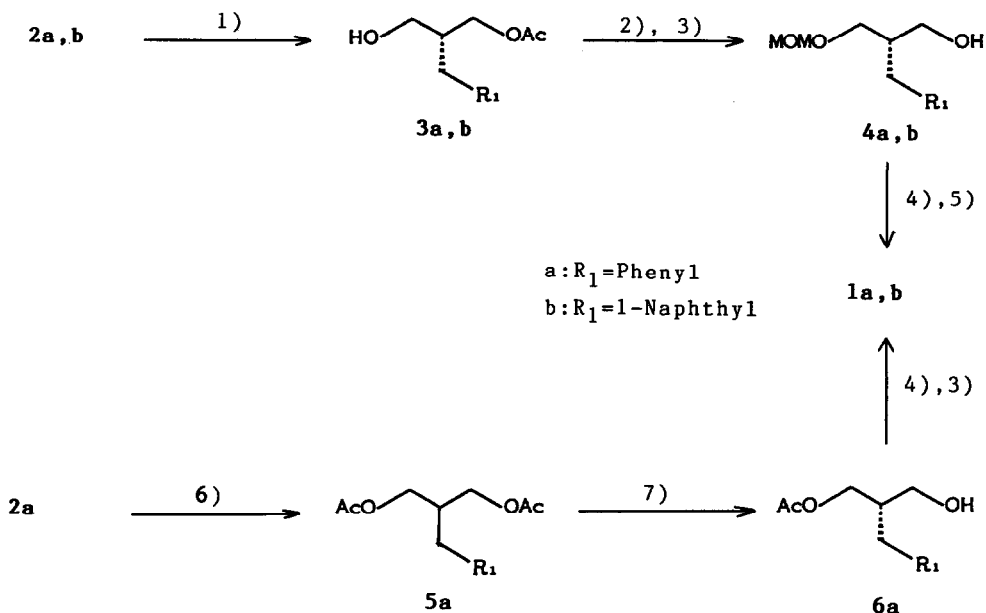


Figure 1



- 1) Lipase P, vinyl acetate 2) MOM-Cl, Et(ⁱPr)₂N, CH₂Cl₂ 3) K₂CO₃, MeOH-H₂O
 4) Jones reagent 5) HCl-H₂O-THF 6) Ac₂O, DMAP, Et₃N, CH₂Cl₂
 7) Lipase P, 0.1 M KPB (pH7.0) -30% Acetone

Scheme 1

Table 1

	Yield (%)	$[\alpha]_D^{20}$ CHCl ₃ (conc.)	e.e. (%)
3a	quant.	+28.6 (1.07)	>94 ^a)
3b	93	+35.7 (1.03)	86 ^a)
4a	99	-18.1 (1.04)	>94 ^b)
4b	91	-20.7 (0.98)	86 ^b)
1a	68	+13.9 (0.97)	>98 ^c)d)
1b	50	+33.1 (0.98)	96 ^c)d)
6a	41	-28.1 (1.01)	>94 ^a)

a) determined by HPLC analysis using a column packed with CHIRALCEL OC (iPrOH/hexane =1/9). b) CHIRALCEL OC (iPrOH/hexane=1/19). c) Senshu Pak. Silica-1151-N (ethyl acetate/hexane=1/1) after coupling with L-norleucine tert-butyl ester. d) after recrystallization.

The synthetic route is outlined in Scheme 1.

After a preliminary screening test to find the desirable enzyme, we selected commercially available Lipase P from *Pseudomonas fluorescens*.⁵ The esterification of 2-substituted-1,3-propanediols **2**⁶ catalyzed by Lipase P using vinyl acetate⁷ as the solvent at room temperature proceeded enantioselectively to afford (2R)-2-substituted-1,3-propanediol-1-acetates **3**⁸ in high yield. Treatment of **3** with chloromethylmethyl ether, followed by hydrolysis with potassium carbonate, provided **4** in excellent yield. After Jones oxidation of **4**, deprotection with hydrochloric acid gave crude 2-substituted-3-hydroxypropionic acids **1**. A single recrystallization yielded **1** in high enantiomeric purity.⁹

The optically pure **1a** was also obtained from the diacetate **5a** via enantioselective hydrolysis using Lipase P. The diol **2a** was allowed to react with acetic anhydride in the presence of 4-dimethylaminopyridine-triethylamine to give the diacetate **5a** quantitatively. The incubation of **5a** with Lipase P in 0.1 M phosphate buffer (containing 30% acetone, pH 7.0) at 30°C gave (2S)-2-benzyl-1,3-propanediol-1-acetate **6a** in 41% yield.¹⁰ The Jones oxidation of **6a** followed by hydrolysis with potassium carbonate yielded **1a** in 64% yield.

Thus, we have established the efficient enantioselective synthesis of 2-substituted-3-hydroxypropionic acids from 2-substituted-1,3-propanediols.

References and Notes

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 - Supplied by Amano Pharmaceutical Co.
 - Compound **2a** was derived from commercially available diethyl benzylmalonate quantitatively. Compound **2b** was synthesized starting from 1-naphthaldehyde and diethyl malonate via diethyl 1-naphthylmalonate in 90% yield.
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 - The absolute configuration was determined by comparison of the measured optical rotations to that of the authentic compound synthesized by the method of D. A. Evans. M. Nakano, S. Atsumi, Y. Koike, S. Tanaka, H. Funabashi, J. Hashimoto, H. Morishima, in preparation.
 - 1a**: mp 67.5–68.5°C (colorless needles from diethyl ether–hexane).
1b: mp 116.5–118.0°C (pale brown powder from benzene).
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